

Figure 6. View of the nanochannels of (a) POROF-1 and (b) POROF-2

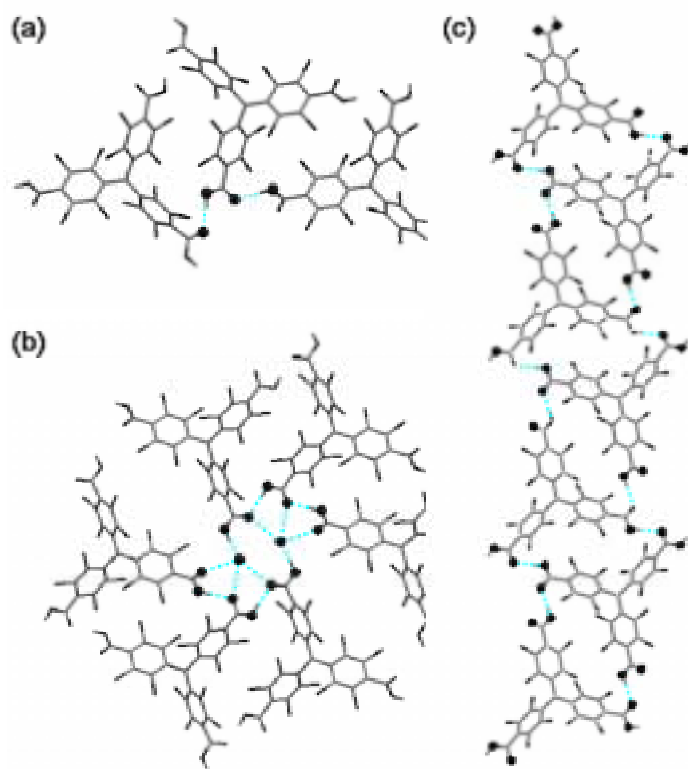


Figure 7. (a) H-bonded trimer of PTMTC radicals; (b) Hexameric motif disrupted by two water molecules; (c) One-dimensional H-bonded chain of PTMTC radicals along the b axis.

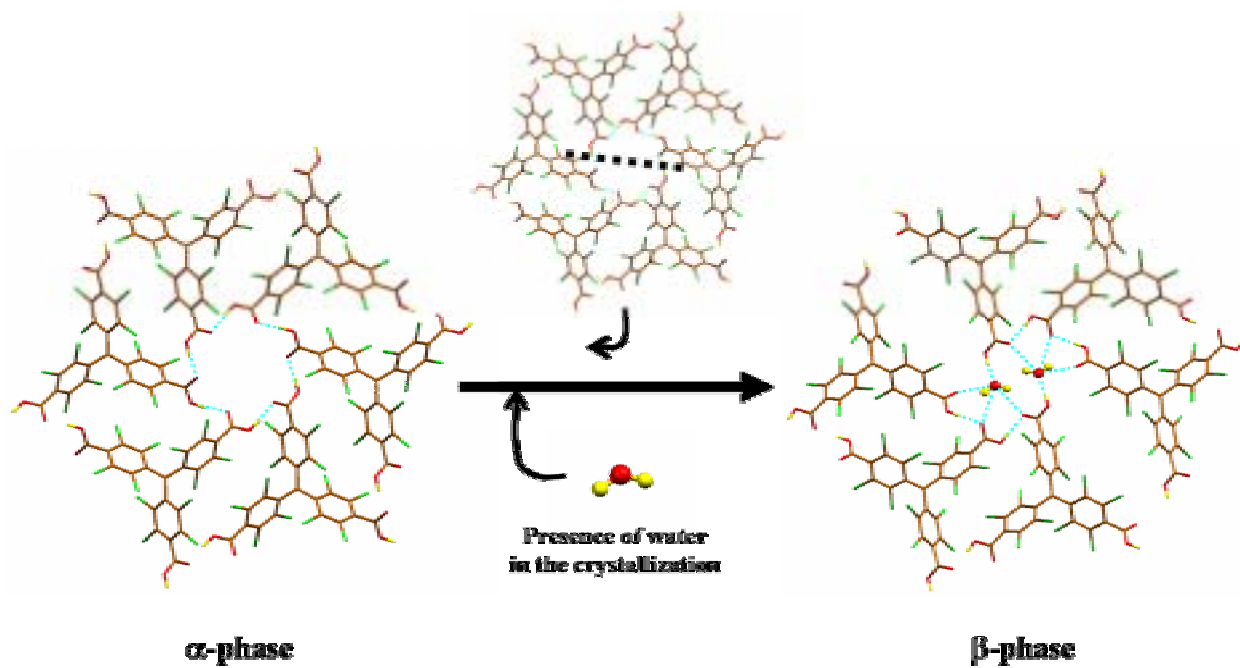


Figure 8. Schematic view of the disruption of the hexameric motif in η -phase due to the presence of water molecules.

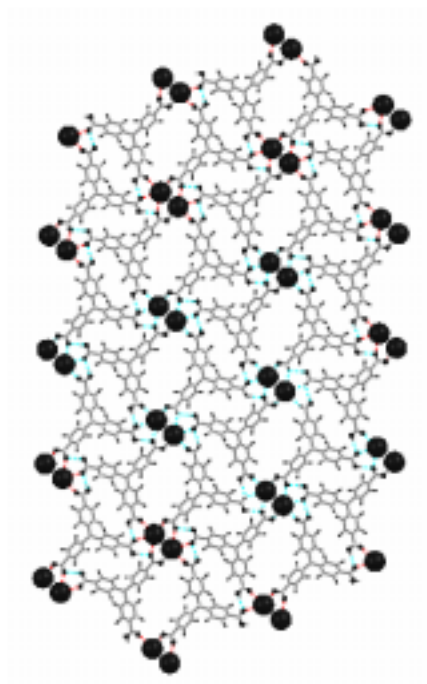


Figure 9. One-dimensional H-bonded chains connected to each other by water molecules along the *bc* plane.

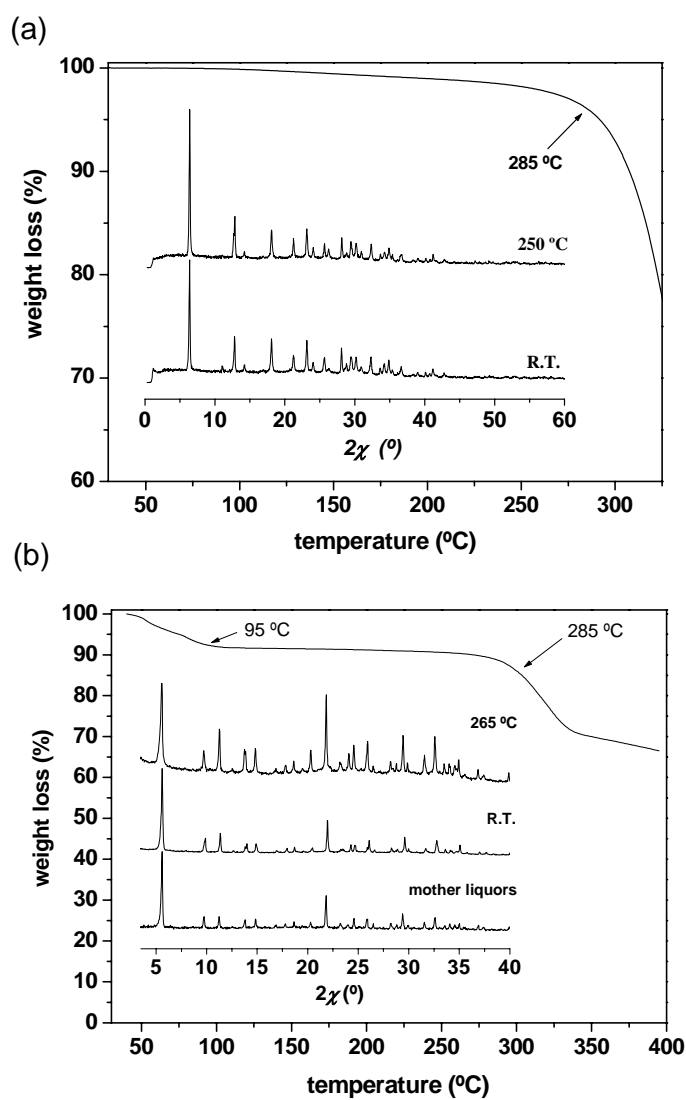


Figure 10. Thermal gravimetric study of a few single crystals of as-synthesized (a) POROF-2 and (b) POROF-1. Inset. Powder X-Ray diffractions of POROF-1 and POROF-2 at different temperatures.

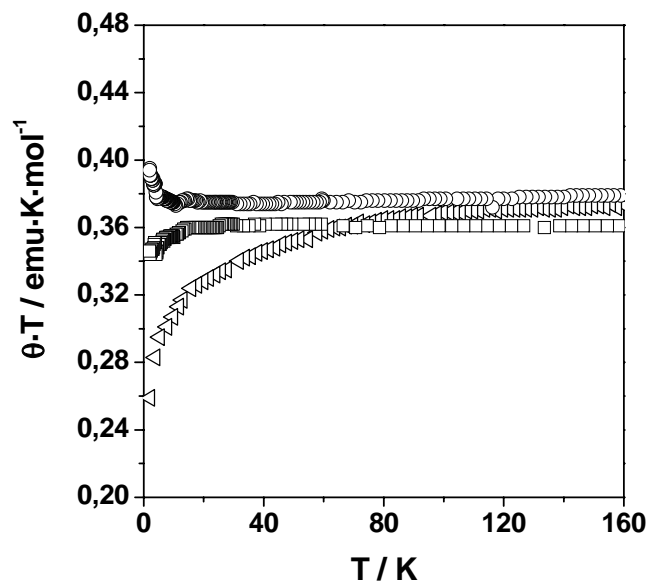


Figure 11. Temperature dependence of the magnetic susceptibility of POROF-1 (∇), POROF-2 (o) and η -phase PTMTC (\square).

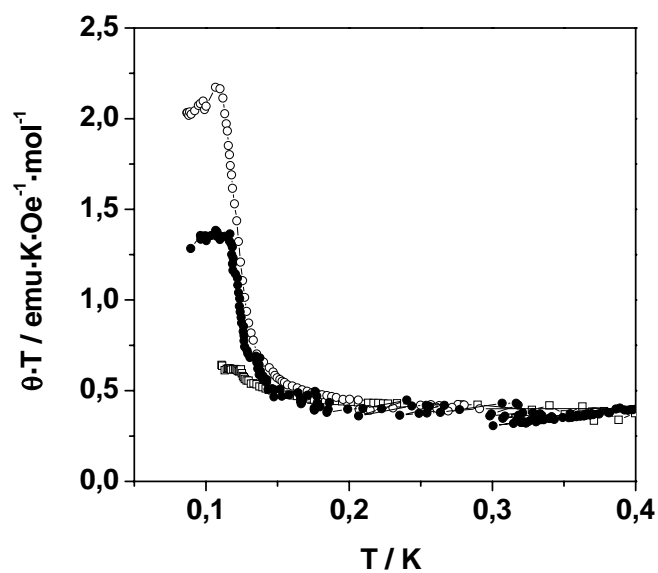


Figure 12. Temperature dependence of the magnetic susceptibility of POROF-2 at lower temperatures at an external magnetic field of 200 (o), 500 (\square) and 1000 (\bullet) Oe.

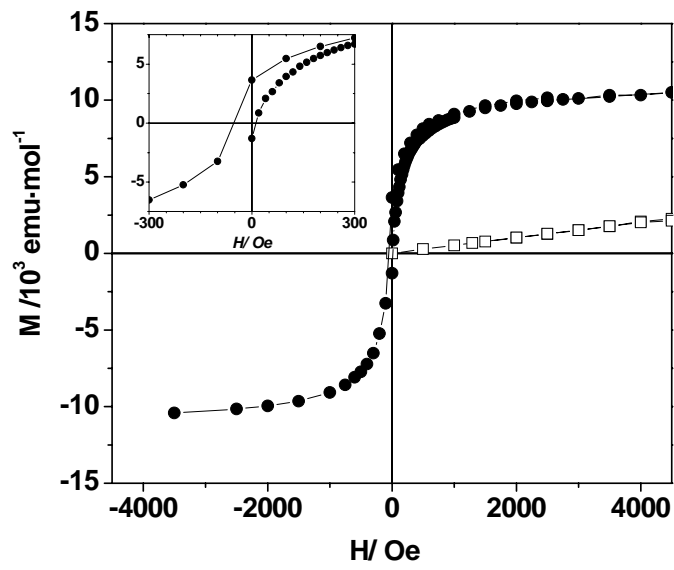
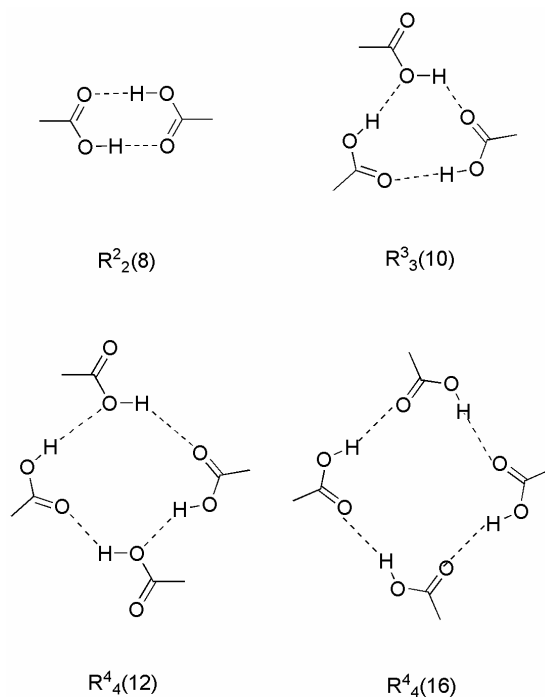


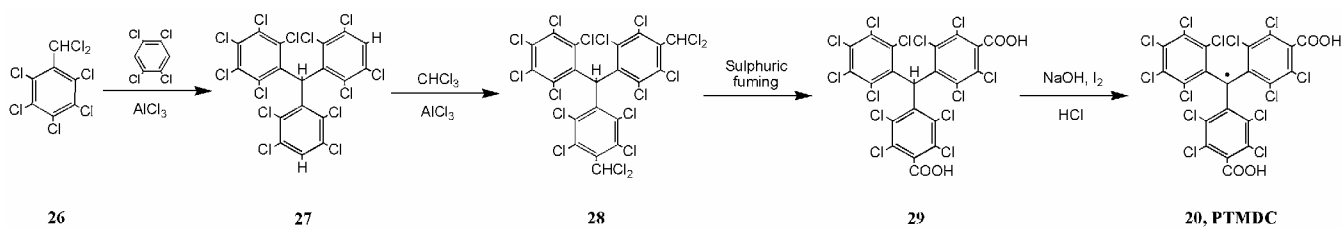
Figure 12. Magnetization curves of POROF-2 at 1350 () and 80 () mK. Inset. Hysteresis curve at 80 mK.

SCHEMES

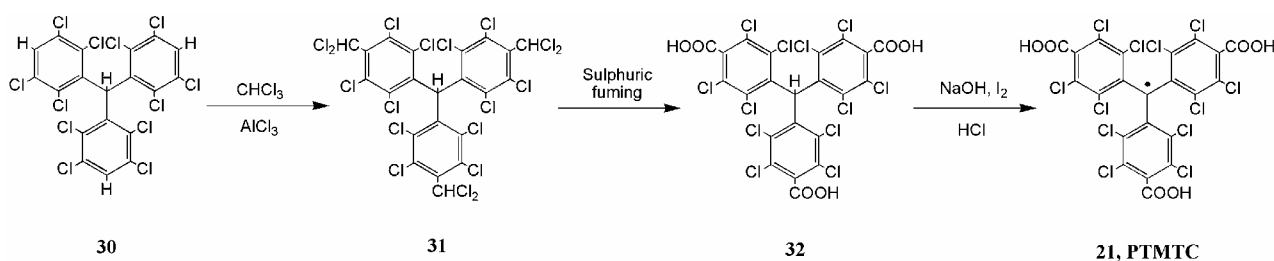
Scheme 1. Cyclic H-Bonded synthons of carboxylic groups.



Scheme 2. Schematic synthesis of PTMDC radical.

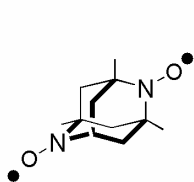
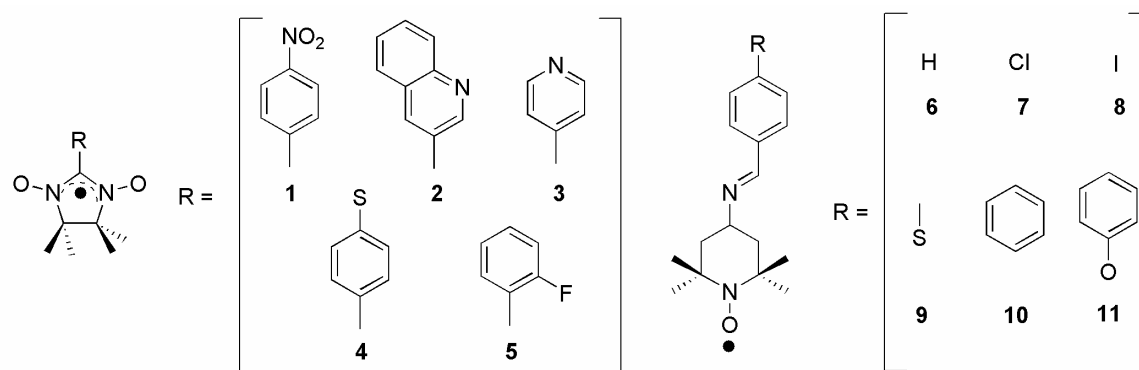


Scheme 3. Schematic synthesis of PTMTC radical.

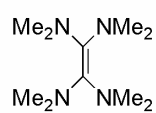


SCHARTS.

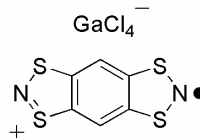
Schart 1



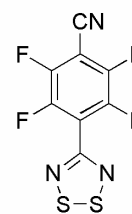
12



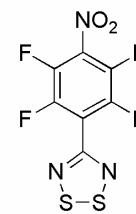
13



14

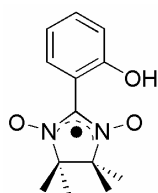


15

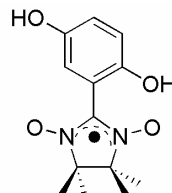


16

Schart 2

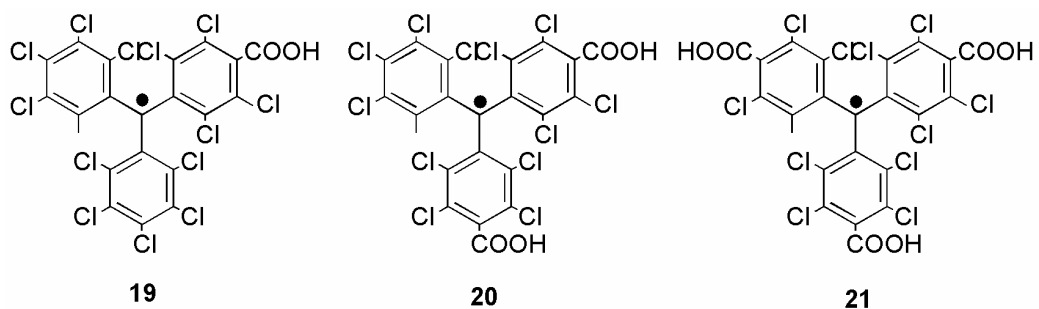


17

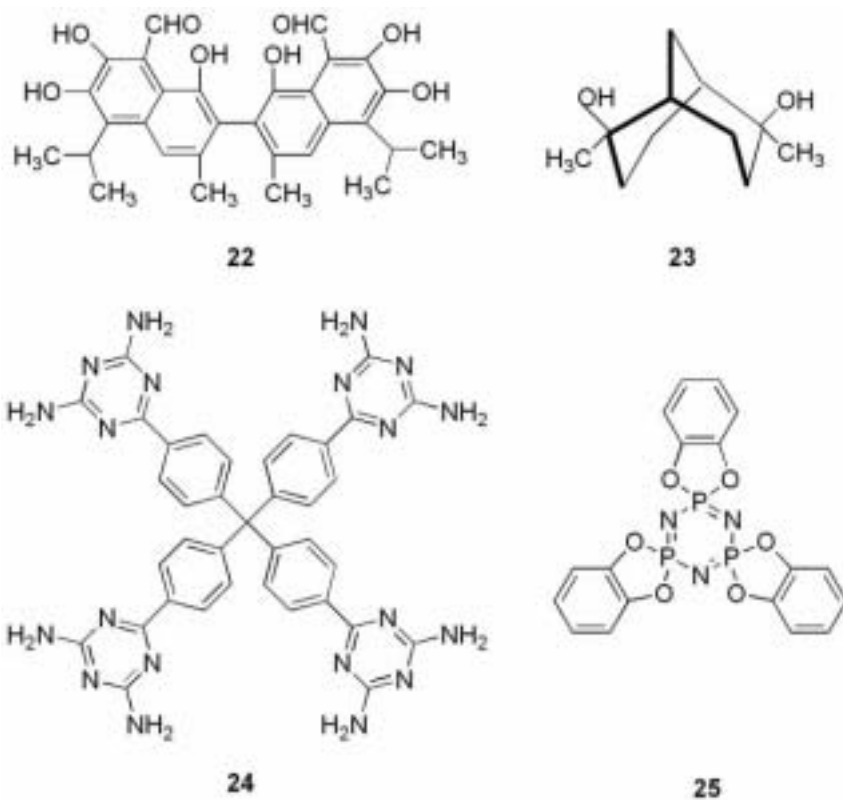


18

Schart 3



Schart 4



TABLES.

Table 1. Crystallographic Data for POROF-1, POROF-2 and η -phase PTMTC.

	POROF-1	POROF-2	η -phase PTMTC
formula	C ₂₇ H ₁₆ Cl ₁₃ O ₄	C ₂₂ H ₃ Cl ₁₂ O ₆	C ₂₃ H ₆ Cl ₁₅ O ₇
fw	865.25	788.64	926.03
cryst syst.	trigonal	trigonal	triclinic
space group	R-3	P-3c1	P-1
a (Å)	31.4651(6)	15.9283(7)	8.7828(6)
b (Å)	31.4651(7)	15.9283(7)	13.4459(8)
c (Å)	18.8447(7)	13.8886(11)	14.172(1)
ζ (°)	90.00	90.00	83.978(4)
η (°)	90.00	90.00	87.636(4)
ν (°)	120.00	120.00	88.648(4)
V(Å ³)	16157.6(8)	3051.6(3)	1662.64(19)
Z	18	4	2
Temp (K)	233(2)	233(2)	223(2)
R ₁	0.0461	0.0585	0.0427
wR ₂	0.1183	0.1418	0.0949
GOF	1.025	1.133	1.049

Table 2. Dihedral angles (°) between the phenyl groups (Ar) and the reference plane and angle between carboxyl group and phenyl rings.

Compound	Ar	Angle (°)	Carboxyl	Angle (°)
PTMDC (POROF-1)	C3-C8	45.2	O1-C1-O2	81
	C10-C15	49.0	O3-C2-O4	90
	C16-C21	43.7		
PTMTC (POROF-2)	C2-C7	49.7	O1-C1-O2	87
PTMTC (η -phase)	C5-C10	51.7	O1-C1-O2	88
	C11-C16	49.4	O3-C2-O4	79
	C17-C22	40.6	O5-C3-O6	81

Table 3. Supramolecular interactions in PTMMC, POROF-1, POROF-2 and η -PTMTC.

	PTMMC	POROF-1	POROF-2	η -phase
Primary structure	Discrete H-bonded dimers	2-D H-bonded sheets	2-D H-bonded sheets	1-D H-bonded chains through water molecules
H-bonded synthon	[R ² ₂ (8)]	[R ² ₂ (8)] and [R ⁶ ₆ (24)]	[R ⁶ ₆ (24)]	[R ⁴ ₄ (12)] and [R ³ ₃ (8)]
D...A H-bonding dist (Å)	O1...O2 = 2.676	O3...O4 = 2.677 O3...Cl11 _p = 3.092 O1...O2 = 2.692	O1...O2 = 2.657	O2...O3 = 2.655 O4...O5 = 2.749 O1...O7 _(water) = 2.534 O2...O7 _(water) = 3.015 O4...O7 _(water) = 2.991 O6...O7 _(water) = 2.908
Cl...Cl contact dist (Å)		Cl1 _m ...Cl4 _m = 3.48 Cl2 _o ...Cl4 _m = 3.32 Cl5 _o ...Cl12 _m = 3.45	Cl2 _o ...Cl3 _o = 3.44	Cl6 _o ...Cl12 _m = 3.45
Secondary structure	H-bonded chains connected by Cl...Cl contacts.	Sheets connected by H-bondings and Cl...Cl contacts.	Sheets connected by Cl...Cl contacts.	Sheets connected by H-bondings and Cl...Cl contacts.
D...A H-bonding dist (Å)	O2...Cl12 _p = 3.263	O4...Cl9 _o = 3.23		O3...Cl6 _o = 3.27
Cl...Cl contact dist (Å)	Cl1 _m ...Cl9 _o = 3.35 Cl2 _o ...Cl8 _m = 3.33 Cl3 _o ...Cl11 _m = 3.22 Cl4 _m ...Cl7 _p = 3.46 Cl5 _o ...Cl5 _o = 3.22 Cl10 _o ...Cl13 _m = 3.40	Cl1 _m ...Cl7 _m = 3.32 Cl2 _o ...Cl7 _m = 3.39 Cl3 _o ...Cl6 _m = 3.49	Cl2 _o ...Cl4 _m = 3.33	Cl4 _m ...Cl10 _o = 3.38 Cl5 _m ...Cl7 _o = 3.33

Table 4. Elemental analysis of POROF-1 and POROF-2

Compound	Formula	Calculated	Found
POROF-2 (R.T.)	C ₂₂ Cl ₁₂ O ₆ H ₃	33.50%C, 0.38%H	33.65%C, 0.32%H
POROF-2 (265°C)	C ₂₂ Cl ₁₂ O ₆ H ₃	33.50%C, 0.38%H	33.80%C, 0.52%H
POROF-1 (R.T.)	C ₂₁ Cl ₁₃ O ₄ H ₂ ·Hexan	36.54%C, 1.56%H	36.10%C, 1.08%H
POROF-1 (250°C)	Ĉ ₂₁ Cl ₁₃ O ₄ H ₂ ·	32.40%C, 0.25%H	32.93%C, 0.42%H

References

- ¹ M. Tamura, Y. Nakazawa, D. Shiomi, K. Nozawa, Y. Hosokoshi, M. Ishikawa, M. Takahashi, M. Kinsohita, *Chem. Phys. Lett.*, **1991**, 186, 401.
- ² For examples of aromatic derivatives of nitronyl nitroxides ferromagnets: (a) Sugano, T., Tamura, M., Kinoshita, M., Sakai, Y., Ohashi, Y., *Chem. Phys. Lett.*, **1992**, 200, 235. (b) Awaga, K., Inabe, T., Maruyama, Y., *Chem. Phys. Lett.*, **1992**, 190, 349. (c) Caneschi, A., Ferraro, F., Gatteschi, D., Le Lirzin, A., Novak, M. A., Rentscher, E., Sessoli, R., *Adv. Mater.*, **1995**, 7, 476. (d) Nakatsuji, S., Saiga, M., Haga, N., Naito, A., Hirayama, T., Nakagawa, M., Oda, Y., Anzai, H., Suzuki, K., Enoki, T., Mito, M., Takeda, K., *New J. Chem.*, **1998**, 22, 275.
- ³ For examples of aromatic derivatives of nitroxides ferromagnets: (a) Nogami, T., Tomioka, K., Ishida, T., Yoshikawa, H., Yasui, M., Iwasaki, F., Iwamura, H., Takeda, N., Ishikawa, M., *Chem. Lett.*, **1994**, 29. (b) Nogami, T., Ishida, T., Tsuboi, H., Yoshikawa, H., Yamamoto, H., Yasui, M., Iwasaki, F., Iwamura, H., Takeda, N., Ishikawa, M., *Chem. Lett.*, **1995**, 635. (c) Togashi, K., Imachi, R., Tomioka, K., Tsuboi, H., Ishida, T., Nogami, T., Takeda, N., Ishikawa, M., *Bull. Chem. Soc. Jpn.*, **1996**, 69, 2821.
- ⁴ Chiarelli, R., Novak, M. A., Rassat, A., Tholance, J. L., *Nature*, **1993**, 363, 147.
- ⁵ Allemand, M. P., Khhermani, K. C., Koch, A., Wudl, F., Holczer, K., Donovan, S., Grúner, G., Thompson, J. D., *Science*, **1991**, 253, 301.
- ⁶ Fujita, W., Awaga, K., *Chem. Phys. Lett.* **2002**, 357, 385.
- ⁷ (a) Banister, A. J., Bricklebank, N., Lavender, I., Rawson, J. M., Gregory, C. I., Tanner, B. K., Clegg, W., Elsegood, R. J., Palacio, F., *Angew. Chem. Int. Ed.*, **1996**, 35, 2533. (b) Alberola, A., Less, R. J., Pask, C. M., Rawson, J. M., Palacio, F., Oliete, P., Paulsen, C., Yamaguchi, A., Farley, R. D., Murphy, D. M., *Angew. Chem. Int. Ed.*, **2003**, 42, 4782.
- ⁸ Maddox, J., *Nature*, **1988**, 335, 201.
- ⁹ Desiraju, G. R. *Crystal Engineering: the Design of Organic Solids*; Elsevier: Amsterdam, 1989.
- ¹⁰ (a) Lehn, J.-M., *Supramolecular Chemistry*, VCH, Weinheim, **1995**. (b) Aakeröy, C. B., Beatty, A. M., Helfrich, B. A., *Angew. Chem. Int. Ed.*, **2001**, 40, 3240.
- ¹¹ (a) Cirujeda, J., Hernández-Gasió, C., Rovira, Stanger, J. L., Turek, P., Veciana, J., *J. Mater. Chem.*, **1995**, 5, 243. (b) Veciana, J., Cirujeda, J., Rovira, C., Vidal-Gancedo, J., *Adv. Mater.*, **1995**, 7, 221.
- ¹² Matsushita, M. M., Izuoka, A., Sugawara, T., Kobayashi, T., Wada, N., Takeda, N., Ishikawa, M., *J. Am. Chem. Soc.*, **1997**, 119, 4369.
- ¹³ Cirujeda, J., Mas, M., Molins, E., Lanfranc de Panthou, F., Laugier, J., Park, J. G., Paulsen, C., Rey, P., Rovira, C., Veciana, J., *J. Chem. Soc., Chem. Commun.*, **1995**, 709.
- ¹⁴ Sugawara, T., Matsushita, M. M., Izuoka, A., Wadam, N., Takeda, N., Ishikawa, M., *J. Chem. Soc., Chem. Commun.*, **1994**, 1723.
- ¹⁵ (a) Hernández, E., Mas, M., Molins, E., Rovira, C., Veciana, J., *Angew. Chem. Int. Ed.*, 1993, 32, 882. (b) Cirujeda, J., Ochando, L. E., Amigó, J. M., Rovira, C., Rius, J., Veciana, J., *Angew. Chem. Int. Ed.*, 1995, 34, 55.
- ¹⁶ Leiserowitz, L., *Acta Cryst.*, **1976**, B32, 775.
- ¹⁷ Kolotuchin, S. V., Fenlon, E. E., Wilson, S. R., Lowth, C. J., Zimmerman, S. C., *Angew. Chem. Int. Ed.*, **1995**, 34, 2654.
- ¹⁸ Ballester, M., *Acc. Chem. Res.*, **1985**, 12, 380.

-
- ¹⁹ Kolotuchin, S. V., Thiessen, P. A., Fenlon, E. E., Wilson, S. R., Loweth, C. J., Zimmerman, S. C., *Chem. Eur. J.*, **1999**, *5*, 2537, and references therein.
- ²⁰ For a general review of the state-of-art of the field: (a) Nangia, A., *Curr. Opin. Solid State Mater Sci.*, **2001**, *5*, 115. (b) Zaworotko, M. J., *Angew. Chem. Int. Ed.*, **2000**, *39*, 3052. (c) Langley, P. J., Hulliger, J., *Chem. Soc. Rev.*, **1999**, *28*, 279. (d) Desiraju, G. R., *Curr. Opin. Solid State Mater Sci.*, **1997**, *2*, 451.
- ²¹ Selected examples of pure open-framework structures: (a) Miyahara, Y., Abe, K., Inazu, T., *Angew. Chem. Int. Ed.*, **2002**, *41*, 3020. (b) Bong, D. T., Ghadiri, M. R., *Angew. Chem. Int. Ed.*, **2001**, *40*, 2163. (c) Sada, K., Sugahara, M., Kato, K., Miyata, M., *J. Am. Chem. Soc.*, **2001**, *123*, 4386. (d) Kiang, Y. H., Lee, S., Xu, Z., Choe, W., Gardner, G. B., *Adv. Mater.*, **2000**, *12*, 767. (e) Müller, T., Hulliger, J., Seichter, W., Weber, E., Weber, T., Wübbenhorst, M., *Chem. Eur. J.*, **2000**, *6*, 54. (f) Kobayashi, K., Shirasaka, T., Sato, A., Horst, E., Furukawa, N., *Angew. Chem. Int. Ed.*, **1999**, *38*, 3483. (g) Biradha, K., Dennis, D., MacKinnon, V. A., Sharma, C. V. K., Zaworotko, M. J., *J. Am. Chem. Soc.*, **1998**, *120*, 11894. (h) Russel, V. C., Evans, C. C., Li, W., Ward, M. D., *Science*, **1997**, *276*, 575
- ²² (a) Yaghi, O. M., O'Keefe, M., Ockwig, N. W., Chae, H. K., Eddaoudi, M., Kim, J., *Nature*, **2003**, *423*, 705. (b) Robson, R., *J. Chem. Soc., Dalton Trans.*, **2000**, 3735 (c) Moulton, B., Zaworotko, M. J., *Chem. Rev.*, **2001**, *101*, 1629. (d) Blake, A.J., Champness, N.R., Hubberstey, P., Schröder M., Withersby, M.A., *Coord. Chem. Rev.*, **1999**, *183*, 117.
- ²³ Eddaoudi, M., Kim, J., Rosi, N., Vodak, D., Wachter, J., O'Keefe, M., Yaghi, O. M., *Science*, **2002**, *295*, 469.
- ²⁴ Ibragimov, B. T., Talipov, S. A., Aripov, T. F., *J. Incl. Phenom. Mol. Rec.*, 1994, *17*, 317.
- ²⁵ Ung, A. T., Gizachew, D., Bishop, R., Scudder, M. L., Dance, I. G., Craig, D. C., *J. Am. Chem. Soc.*, **1995**, *117*, 8745.
- ²⁶ Brunet, P., Simard, M., Wuest, J. D., *J. Am. Chem. Soc.*, **1997**, *119*, 2737.
- ²⁷ Sozzani, P., Comotti, A., Simonutti, R., Meersmann, T., Logan, J. W., Pines, A., *Angew. Chem. Int. Ed.*, **2000**, *39*, 2695.
- ²⁸ Maspoch, D., Gerbier, P., Catala, L., Vidal-Gancedo, J., Wurst, K., Rovira, C., Veciana, J. *Chem. Eur. J.* **2002**, *8*, 3635.
- ²⁹ Sheldrick, G. M.: SHELXL-97, Program for Crystal Structure refinement, University of Göttingen, Germany **1997**.
- ³⁰ Domingo, V. M., Castañer, J., Riera, J., Labarta, A., *J. Org. Chem.*, **1994**, *59*, 2604.
- ³¹ Ballester, M., Riera, J., Castañer, J., Rovira, C., Armet, O., *Synthesis* **1986**, 64.
- ³² (a) Armet, O., Veciana, J., Rovira, C., Riera, J., Castañer, J., Molins, E., Rius, J., Miravittles, C., Olivella, S., Brichfeus, J., *J. Phys. Chem.*, **1987**, *91*, 5608. (b) Sedó, J., Ventosa, N., Molins, M^aA., Pons, M., Rovira, C., Veciana, J., *J. Org. Chem.*, **2001**, *66*, 1567.
- ³³ Moorthy, J. N., Natajaran, R., Mal, P., Venugopalan, P., *J. Am. Chem. Soc.*, **2002**, *124*, 6530.